| C3 | $0.1283(4)$ | $0.0934(2)$ | $-0.0941(5)$ | $0.051(1)$ |
| :--- | ---: | :---: | :---: | :---: |
| C4 | $-0.2889(7)$ | $0.1775(4)$ | $0.0221(11)$ | $0.089(1)$ |
| C5 | $0.3413(5)$ | $0.1576(4)$ | $0.2350(8)$ | $0.077(1)$ |
| C6 | $0.2480(7)$ | $0.0376(4)$ | $-0.3189(8)$ | $0.080(1)$ |

Table 4. Selected geometric parameters ( $\left(\begin{array}{l} \\ ,\end{array}{ }^{\circ}\right.$ ) for (III)

| $\mathrm{O} 1-\mathrm{Cl} 1$ | $1.218(4)$ | $\mathrm{N} 2-\mathrm{C} 2$ | $1.370(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 2-\mathrm{C} 2$ | $1.213(4)$ | $\mathrm{N} 2-\mathrm{C} 1$ | $1.392(5)$ |
| $\mathrm{O} 3-\mathrm{C} 3$ | $1.23(4)$ | $\mathrm{N} 2-\mathrm{C} 4$ | $1.464(5)$ |
| $\mathrm{O} 3-\mathrm{C} 6$ | $1.453(5)$ | $\mathrm{N} 3-\mathrm{C} 3$ | $1.364(5)$ |
| $\mathrm{N} 1-\mathrm{C} 3$ | $1.284(4)$ | $\mathrm{N} 3-\mathrm{C} 2$ | $1.382(4)$ |
| $\mathrm{N} 1-\mathrm{C} 1$ | $1.365(5)$ | $\mathrm{N} 3-\mathrm{C} 5$ | $1.452(5)$ |
| $\mathrm{C} 3-\mathrm{O} 3-\mathrm{C} 6$ | $116.5(3)$ | $\mathrm{O} 1-\mathrm{Cl}-\mathrm{N} 2$ | $119.5(3)$ |
| $\mathrm{C} 3-\mathrm{N} 1-\mathrm{C} 1$ | $118.2(3)$ | $\mathrm{N} 1-\mathrm{Cl}-\mathrm{N} 2$ | 18.0 |
| $\mathrm{C} 2-\mathrm{N} 2-\mathrm{Cl} 1$ | $123.4(3)$ | $\mathrm{O} 2-\mathrm{C} 2-\mathrm{N} 2$ | $123.3(3)$ |
| $\mathrm{C} 2-\mathrm{N} 2-\mathrm{C} 4$ | $117.0(4)$ | $\mathrm{O} 2-\mathrm{C} 2-\mathrm{N} 3$ | $121.1(3)$ |
| $\mathrm{C} 1-\mathrm{N} 2-\mathrm{C} 4$ | $119.4(4)$ | $\mathrm{N} 2-\mathrm{C} 2-\mathrm{N} 3$ | $15.6(3)$ |
| $\mathrm{C} 3-\mathrm{N} 3-\mathrm{C} 2$ | $118.6(3)$ | $\mathrm{N} 1-\mathrm{C} 3-\mathrm{O} 3$ | $121.7(3)$ |
| $\mathrm{C} 3-\mathrm{N} 3-\mathrm{C} 5$ | $122.8(3)$ | $\mathrm{N} 1-\mathrm{C} 3-\mathrm{N} 3$ | $126.1(3)$ |
| $\mathrm{C} 2-\mathrm{N} 3-\mathrm{C} 5$ | $118.4(3)$ | $\mathrm{O} 3-\mathrm{C} 3-\mathrm{N} 3$ | $12.2(3)$ |
| $\mathrm{O} 1-\mathrm{Cl}-\mathrm{N} 1$ | $122.5(4)$ |  |  |

For both compounds, program(s) used to solve structures: SHELXS86 (Sheldrick, 1985); program(s) used to refine structures: SHELXL93 (Sheldrick, 1993).

This research was partially supported by Technion V. P. R. Fund - E. and M. Mendelson Research Fund, and by the Fund for the Promotion of Research at the Technion.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1180). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Acta Cryst. (1995). C51, 2424-2426

# 5,5-Dimethyl-1,3-dioxacyclohexane-2-spiro( $2^{\prime}, \gamma$-dimethyl- $3^{\prime}$-oxo- $\mathbf{2}^{\prime}$-cyclohexaneacrylic Acid) 

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## Abstract

The cyclohexanone ring in the title compound, $\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{5}$, adopts a chair conformation with the methyl and acrylic acid groups in axial and equatorial positions, respectively. The dihedral angle between the planes of the two rings is $73(1)^{\circ}$. The bond angles around the spiro C atom are all nearly tetrahedral [104.8 (2)$113.0(2)^{\circ}{ }^{\circ}$.

## Comment

With the aim of elucidating the influence of the spiro $C$ atom on ring-skeleton conformation, we have examined the conformation of several cyclohexanone-2methylacrylic acid derivatives; the results of the X-ray analysis of one of these derivatives, (I), is reported here.

(I)

The two mean ring planes form a dihedral angle of $73^{\circ}$; the least-squares planes calculation indicates that the two rings are not individually planar. Following the Cremer \& Pople (1975) notation, the ringpuckering parameters for the cyclohexanone ring are $Q_{2}$
$=0.155$ (3) $\AA$ and $\theta_{2}=+78.1(2)^{\circ}$. The most unusual feature of the angles is the pattern around $\mathrm{C}(10)$, the spiro $C$ atom. The six angles around $C(10)$ are 110.5 (2), 106.6 (2), 111.4 (2), 104.8 (2), 110.3 (2) and $113.0(2)^{\circ}$. Of the three larger angles, two are endocyclic: $\mathrm{O}(4)-$ $\mathrm{C}(10)-\mathrm{O}(5)$ and $\mathrm{O}(5)-\mathrm{C}(10)-\mathrm{C}(9)$ with values of 110.5 (2) and $110.3(2)^{\circ}$, respectively. The magnitudes of these angles are obviously greater than the normal $s p^{3}$ value as a result of the strain of bonding to the O atoms within the compressed ring system. The other angle of the three largest is $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(5)$ with a value of $113.0(2)^{\circ}$; this is the result of the strain of the methyl and acrylic acid groups bonded at the $\mathrm{C}(5)$ atom. The two ring O atoms are in an axial positions (with respect to the opposite ring). The smallest angle around the spiro C atom is $\mathrm{O}(4)-\mathrm{C}(10)-\mathrm{C}(5)$ with a value of $106.6(2)^{\circ}$; this is due to the attraction between the ring O atoms and the H atoms of the methyl group on the opposite ring. Atoms $\mathrm{C}(15)$ and $\mathrm{C}(16)$ are equatorial and axial, respectively.

In the acrylic acid group $\{\mathrm{O}(1)-\mathrm{C}(1)[-\mathrm{O}(2)]-$ $\mathrm{C}(2)\}$, the delocalization is well distributed over the three bonds. Consequently, the $\mathrm{O}(2)-\mathrm{C}(1)$ distance of 1.224 (3) $\AA$ is nearer to those observed in aldehydes and ketones (Sutton, 1965) and in $\alpha$-oxalic acid (Derissen \& Smit, 1974). The C(1)-C(2) bond length is in agreement with the values observed in the two analogous derivatives analysed by Koer, De Kok \& Romers (1972), and by Haltiwanger, Walba \& Wand (1980).


Fig. 1. View of the title compound with displacement ellipsoids plotted at the $50 \%$ probability level.

## Experimental

Crystal data
$\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{5}$
$M_{r}=296.36$
Monoclinic
$P 2_{1} / a$
$a=13.049$ (1) $\AA$
$b=10.411$ (1) $\AA$
$c=12.317(3) \AA$
$\beta=105.5(1)^{\circ}$
$V=1612.8 \AA^{3}$
$Z=4$
$0.35 \times 0.28 \times 0.12 \mathrm{~mm}$ Colourless
$D_{x}=1.22 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}=1.25 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ measured by flotation

## Data collection

CAD-4 diffractometer $\omega$ scans
Absorption correction:
none
3154 measured reflections
2831 independent reflections
1641 observed reflections
$[I>3 \sigma(I)]$

## Refinement

Refinement on $F$
$R=0.050$
$w R=0.062$
1641 reflections
287 parameters
H atoms refined isotropically
$w=1 /\left[\sigma^{2}(F)+0.0004 F^{2}\right]$

$$
\begin{aligned}
& (\Delta / \sigma)_{\max }=0.56 \\
& \Delta \rho_{\max }=0.16(3) \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=0.08(3) \mathrm{e} \AA^{-3} \\
& \text { Extinction correction: none } \\
& \text { Atomic scattering factors } \\
& \text { from Cromer \& Waber } \\
& \text { (1974) }
\end{aligned}
$$

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\AA^{2}$ )

$$
B_{\mathrm{eq}}=\left(8 \pi^{2} / 3\right) \Sigma_{i} \Sigma_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}
$$

|  | $x$ | $y$ | $z$ | $B_{\text {eq }}$ |
| :---: | :---: | :---: | :---: | :---: |
| 01 | 0.3798 (2) | 0.5571 (2) | 0.0200 (2) | 5.65 (5) |
| O 2 | 0.5010 (1) | 0.6487 (2) | -0.0483 (2) | 4.33 (4) |
| 03 | 0.3039 (2) | 0.7030 (2) | -0.2995 (2) | 5.88 (6) |
| O4 | 0.3802 (1) | 1.1319 (2) | -0.1896 (1) | 3.38 (4) |
| O5 | 0.3352 (1) | 1.0401 (2) | -0.3693 (1) | 3.48 (4) |
| Cl | 0.4175 (2) | 0.6553 (3) | -0.0211 (2) | 3.28 (6) |
| C2 | 0.3576 (2) | 0.7779 (2) | -0.0286 (2) | 3.32 (6) |
| C3 | 0.2801 (3) | 0.7858 (3) | 0.0214 (3) | 5.19 (8) |
| C4 | 0.3970 (2) | 0.8907 (3) | -0.0838 (2) | 3.25 (6) |
| C5 | 0.3457 (2) | 0.9074 (2) | -0.2125 (2) | 2.85 (6) |
| C6 | 0.3692 (2) | 0.7874 (3) | -0.2740 (2) | 3.61 (6) |
| C7 | 0.4780 (3) | 0.7723 (3) | -0.2939 (3) | 4.74 (7) |
| C8 | 0.5262 (2) | 0.8942 (3) | -0.3259 (3) | 4.81 (7) |
| C9 | 0.5131 (2) | 1.0059 (3) | -0.2497 (3) | 3.98 (7) |
| C10 | 0.3955 (2) | 1.0248 (2) | -0.2557 (2) | 2.91 (6) |
| C11 | 0.2261 (2) | 0.9244 (3) | -0.2343 (2) | 4.00 (7) |
| C12 | 0.4013 (2) | 1.2545 (3) | -0.2339 (2) | 4.19 (7) |
| C13 | 0.3386 (2) | 1.2741 (3) | -0.3551 (2) | 4.40 (7) |
| C14 | 0.3557 (2) | 1.1565 (3) | -0.4216 (2) | 4.16 (7) |
| C15 | 0.3798 (3) | 1.3927 (3) | -0.4039 (3) | 6.9 (1) |
| C16 | 0.2198 (3) | 1.2886 (4) | -0.3624 (3) | 7.2 (1) |

Table 2. Selected geometric parameters ( $\AA,{ }^{\circ}$ )

| $\mathrm{O} 1-\mathrm{Cl}$ | $1.294(3)$ | $\mathrm{C} 5-\mathrm{C} 6$ | $1.534(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 2-\mathrm{C} 1$ | $1.224(3)$ | $\mathrm{C} 5-\mathrm{C} 10$ | $1.544(4)$ |
| $\mathrm{O} 3-\mathrm{C} 6$ | $1.206(3)$ | $\mathrm{C} 5-\mathrm{C} 11$ | $1.522(4)$ |
| $\mathrm{O} 4-\mathrm{C} 10$ | $1.425(3)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.512(5)$ |
| $\mathrm{O} 4-\mathrm{C} 12$ | $1.443(3)$ | $\mathrm{C} 7-\mathrm{C} 8$ | $1.514(5)$ |
| $\mathrm{O}-\mathrm{C} 10$ | $1.420(3)$ | $\mathrm{C} 8-\mathrm{C} 9$ | $1.532(4)$ |
| $\mathrm{O} 5-\mathrm{C} 14$ | $1.431(4)$ | $\mathrm{C} 9-\mathrm{C} 10$ | $1.530(4)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.487(4)$ | $\mathrm{C} 12-\mathrm{C} 13$ | $1.513(4)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.319(5)$ | $\mathrm{C} 13-\mathrm{C} 14$ | $1.522(4)$ |
| $\mathrm{C} 2-\mathrm{C} 4$ | $1.514(4)$ | $\mathrm{C} 13-\mathrm{C} 15$ | $1.532(5)$ |
| $\mathrm{C} 4-\mathrm{C} 5$ | $1.557(3)$ | $\mathrm{C} 13-\mathrm{C} 16$ | $1.536(5)$ |

$$
\mathrm{C}_{16} \mathrm{H}_{24} \mathrm{O}_{5}
$$

| $\mathrm{C} 10-\mathrm{O} 4-\mathrm{C} 12$ | $113.9(2)$ | $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $115.3(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 10-\mathrm{O} 5-\mathrm{C} 14$ | $114.7(2)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ | $110.8(3)$ |
| $\mathrm{O} 1-\mathrm{Cl}-\mathrm{O} 2$ | $121.4(2)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $110.0(2)$ |
| $\mathrm{O}-\mathrm{Cl}-\mathrm{C} 2$ | $117.0(2)$ | $\mathrm{O} 4-\mathrm{C} 10-\mathrm{O} 5$ | $110.5(2)$ |
| $\mathrm{O} 2-\mathrm{Cl}-\mathrm{C} 2$ | $121.5(2)$ | $\mathrm{O} 4-\mathrm{C} 10-\mathrm{C} 5$ | $106.6(2)$ |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | $118.7(3)$ | $\mathrm{O} 4-\mathrm{C} 10-\mathrm{C} 9$ | $111.4(2)$ |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 4$ | $117.2(2)$ | O5-C10-C5 | $104.8(2)$ |
| $\mathrm{C} 3-\mathrm{C} 2-\mathrm{C} 4$ | $123.8(3)$ | O5-C10-C9 | $110.3(2)$ |
| $\mathrm{C} 2-\mathrm{C} 4-\mathrm{C} 5$ | $116.0(2)$ | $\mathrm{C} 5-\mathrm{C} 10-\mathrm{C} 9$ | $113.0(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 6$ | $108.9(2)$ | O4-C12-C13 | $112.6(2)$ |
| $\mathrm{C} 4-\mathrm{C}-\mathrm{C} 10$ | $109.5(2)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 14$ | $107.7(2)$ |
| $\mathrm{C} 4-\mathrm{C} 5-\mathrm{C} 11$ | $109.6(2)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 15$ | $110.0(2)$ |
| $\mathrm{C} 6-\mathrm{C}-\mathrm{C} 10$ | $108.5(2)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 16$ | $110.0(3)$ |
| $\mathrm{C} 6-\mathrm{C} 5-\mathrm{C} 11$ | $109.8(2)$ | $\mathrm{C} 14-\mathrm{C} 13-\mathrm{C} 15$ | $108.7(3)$ |
| $\mathrm{C} 10-\mathrm{C} 5-\mathrm{C} 11$ | $110.6(2)$ | $\mathrm{C} 14-\mathrm{C} 13-\mathrm{C} 16$ | $109.7(3)$ |
| $\mathrm{O} 3-\mathrm{C} 6-\mathrm{C} 5$ | $120.1(3)$ | $\mathrm{C} 15-\mathrm{C} 13-\mathrm{C} 16$ | $110.6(3)$ |
| $\mathrm{O} 3-\mathrm{C} 6-\mathrm{C} 7$ | $120.7(3)$ | $\mathrm{O} 7-\mathrm{C} 14-\mathrm{C} 13$ | $111.6(2)$ |
| $\mathrm{C} 5-\mathrm{C} 6-\mathrm{C} 7$ | $119.0(2)$ |  |  |

A total of 20 atoms were located from an $E$ map. The remaining atoms were located in succeeding difference Fourier syntheses. H atoms were located and their positions and isotropic displacement parameters were refined. The structure was refined by full-matrix least squares.

Data collection and cell refinement were performed using CAD-4 Software (Enraf-Nonius, 1989). The structure was solved by direct methods (SIR88; Burla et al., 1989). All calculations were performed on a VAX computer using MolEN (Fair, 1990). Molecular graphics were obtained using ORTEP (Johnson, 1965).

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: CR1184). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Acta Cryst. (1995). C51, 2426-2428
Methyl 6-Benzoyl-3,5-diphenyl-1,4-thio-morpholine-2-carboxylate 1,1-Dioxide

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## Abstract

The crystal structure determination of $\mathrm{C}_{25} \mathrm{H}_{23} \mathrm{NO}_{5} \mathrm{~S}$ reveals that the thiomorpholine ring adopts a slightly distorted chair conformation. All the substituents on heterorings are equatorially oriented. There are no short intermolecular distances and the structure is stabilized with van der Waals interactions.

## Comment

Thiomorpholine derivatives have attracted considerable interest because of their various chemotherapeutic properties, such as the antihistaminic activity of promethazine, the psychotherapeutic action of chloropromazine and the insecticidal and anthelmentic properties of prothiazine (Katritzky \& Boulton, 1968). Since the chemical and physiological characteristics of these compounds are directly defined by the conformation of their molecules and their relevant structural data are limited, X-ray crystallographic studies on a series of thiomorpholine derivatives have been undertaken. Here, we report the crystal and molecular structure of the title compound, (I).

(I)

